

F325 June 2014 Unofficial Mark Scheme

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F325 Mark Scheme June 2014 4 1 (a) (ii) (The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound from its gaseous ions (under standard conditions) Award marks as follows. 1st mark: formation of compound from gaseous ions 2nd mark: one mole for compound only DO NOT ALLOW 2nd mark without 1st mark

Oxford Cambridge and RSA

F325 Mark Scheme June 2014 4 1 (a) (ii) (The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound from its gaseous ions (under standard conditions) 99 Award marks as follows.

GCE Chemistry A - The Exam Formula

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F325 Mark Scheme June 2012 Question Answer Marks Guidance
1 (d) For FIRST TWO marking points, assume that the following refer to 'ions', Mg. 2+, etc. For 'ions', ALLOW

Advanced GCE Unit F325: Equilibria, Energetics and Elements

F325 Mark Scheme June 2013 7 Question Answer Marks
Guidance 2 (b) (ii) $k =$ answer to (b)(i) 3.45 units: s⁻¹
Independent mark 2 From 0.045, $k = 0.045$ 3.45 = 0.013 ALLOW
concentration range 3.4–3.5 ALLOW use of unrounded calculator
answer from (b)(i) even if different from answer given on (b)(i)
answer line

Advanced GCE Unit F325: Equilibria, Energetics and Elements

F325 Mark Scheme June 2010 2 Question Expected Answers
Marks Additional Guidance 2 a BrO₃⁻ + 5Br⁻ + 6H⁺ 3Br₂ +
3H₂O 1 ALLOW multiples b graph: Straight/diagonal line through
origin OR 0,0 AND 1st order with respect to BrO₃⁻ 1
ANNOTATIONS MUST BE USED Both explanation and 1st order
required for mark

Mark Scheme for June 2010 - A Level Chemistry

F325 Mark Scheme June 2011 1 Question Answer Mark Guidance
1 (a) (The enthalpy change that accompanies) the formation of
one mole of a(n ionic) compound from its gaseous ions (under
standard conditions) 2 IGNORE 'Energy needed' OR 'energy
required' ALLOW as alternative for compound: lattice, crystal,
substance, solid Note: 1st mark requires 1 mole

F325 Mark Scheme June 2011 Question Answer Mark Guidance 1 ...

F325 Mark scheme June 2016 6 Question Answer Marks
Guidance 1 (b) (ii) 3 FULL ANNOTATIONS MUST BE USED ----- 2
marks: one for each correct isomer TAKE CARE: structures may
be in different orientations and in different order IF BOTH

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isomers are 'correct', but O connectivity wrong, AWARD 1 mark for both structures ...

Mark Scheme for June 2016 - PMT

F325 Mark Scheme January 2011 Question . er . AnswMark . Guidance . 2 (g) 1,1-cyclobutanedicarboxylate ion . O ...

Advanced GCE Unit F325: Equilibria, Energetics and Elements

OCR A-level Chemistry (A) Past Papers and Mark Schemes (H034 & H434) Click on a paper to view then on the purple source button at the bottom right hand corner to download it.
P.J.McCormack

OCR A-level Chemistry (A) Past Papers and Mark Schemes

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F325 Mark Scheme June 2015 Question Answer Marks Guidance 2 (a) NOTE: First 3 marks are ONLY available from an expression : 2 using [NO] Units are marked independently ----- Using values ON THE CURVE in CORRECT expression 1 mark Use of any two correct values for rate and [NO] from graph e.g. for $5.0 \times 10 \dots$

Mark Scheme for June 2015 - PapaCambridge

You can find the paper and official mark scheme here: ... OCR Unit 5 F325 June 2013 Past Paper Work through ... F324 Rings Polymers and Analysis June 2014 Q1- 2a - Duration: 14:39.
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F325 Jun14 Q2a

F325 Mark Scheme January 2013 6 Question Answer Marks Guidance 2 (b) (ii) FIRST, CHECK THE ANSWER ON ANSWER LINE IF answer = $-57.5 \text{ (kJ mol}^{-1}\text{)}$ award 3 marks energy change = $70.0 \times 4.18 \times 16.5$

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(March 2014) (Learn how and when to remove this template message) Mark Rosewater. Born May 25, 1967 (age 53)

Occupation: Head designer for Magic: The Gathering: Mark Rosewater (born May 25, 1967) is the head designer for Magic: The Gathering, a position he has held since 2003. Biography. Rosewater grew up in Pepper Pike ...

Mark Rosewater - Wikipedia

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